

On the array of 1 at.% Ce-doped ZnO nanorod prepared by electrochemical deposition with their structure and characterization

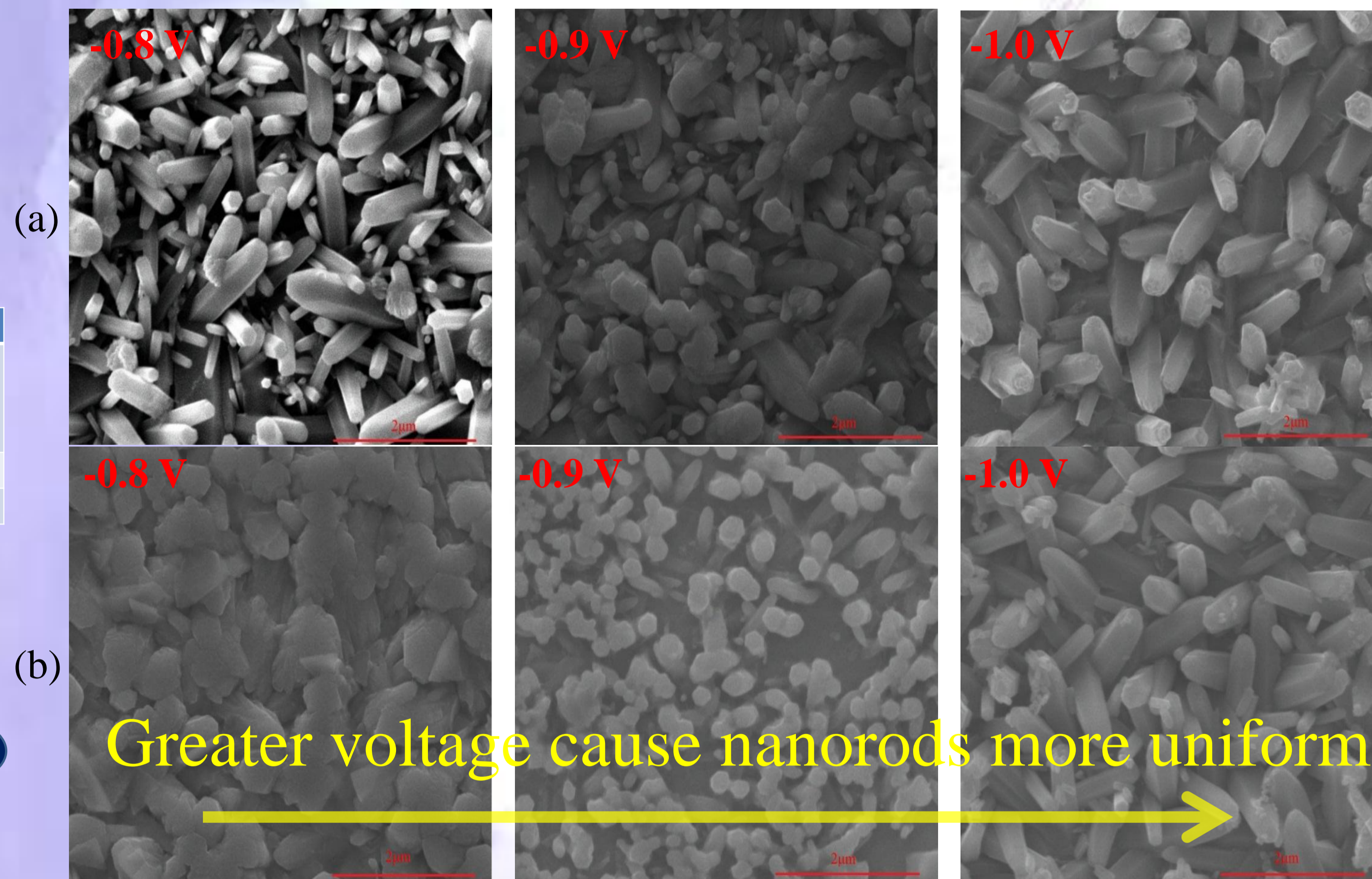
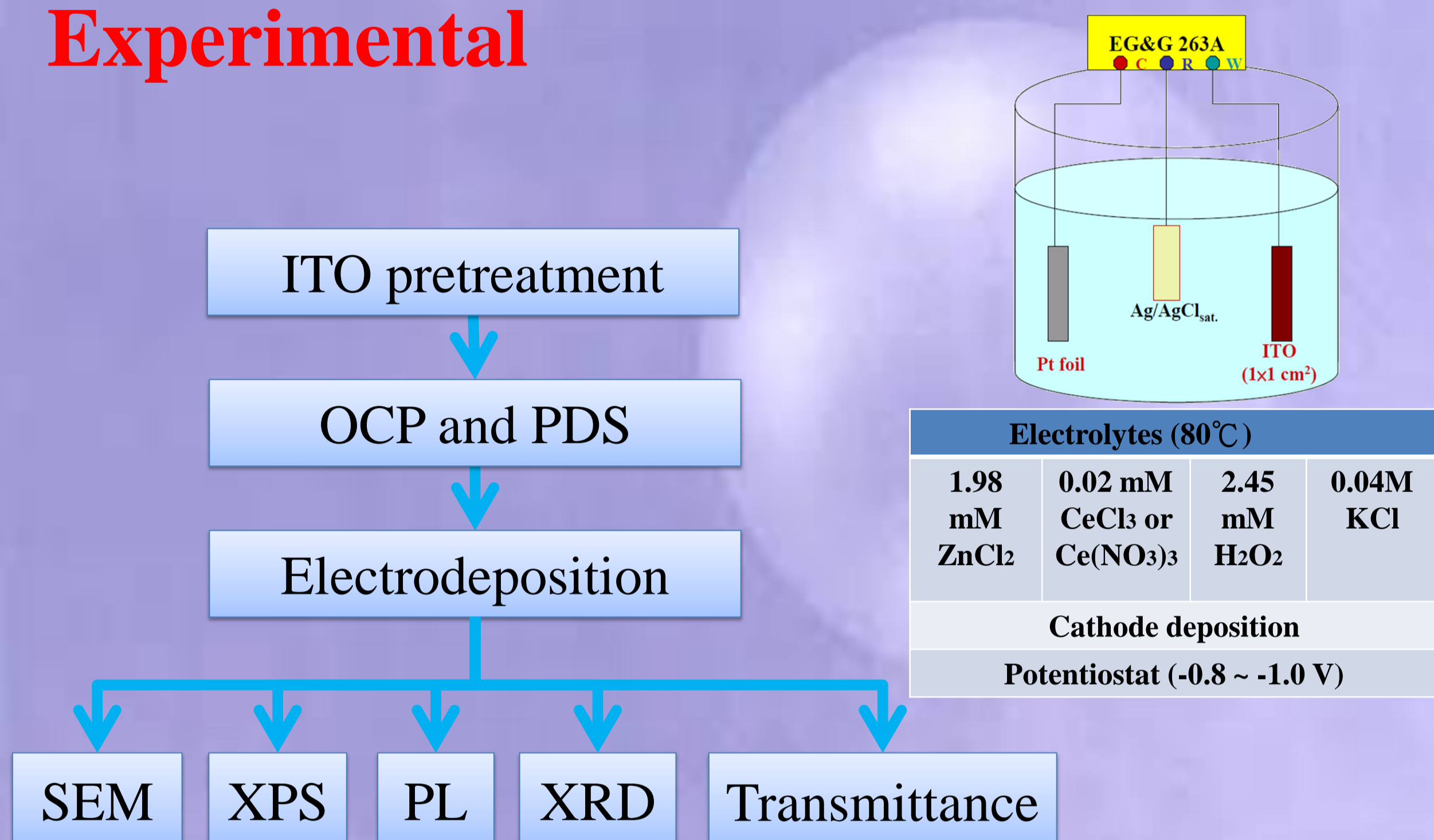
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Abstract

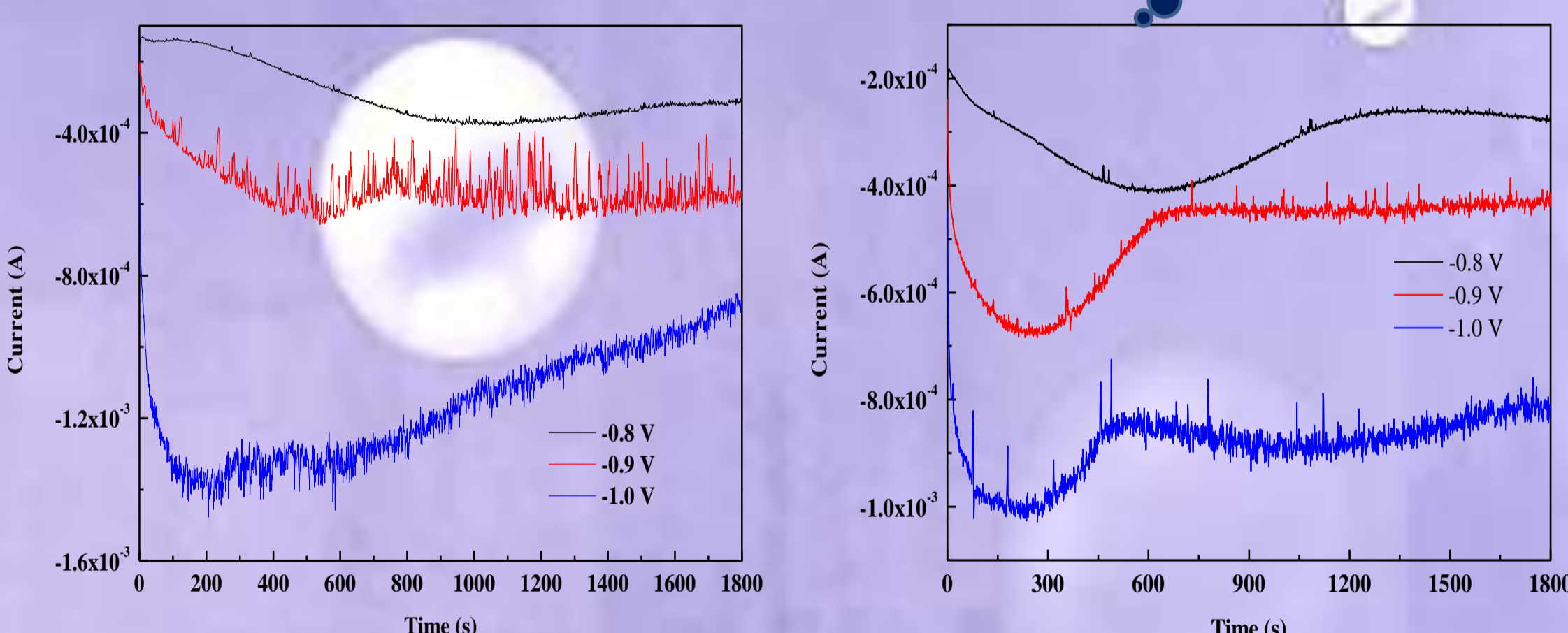
ZnO-based semiconductor is a transparent material for many applications, such as some electronic devices, light emitting diode, fluorescent material, sensor, and solar cell. In this study, ZnO-nanorods were doped with 1 at.% Ce on ITO substrate by electrodeposition. We prepared our electrolyte from ZnCl₂ and H₂O₂, and 0.04M KCl. Our purpose is to improve ZnO's transmittance, and compare electrochemical curve with different precursor, CeCl₃ and Ce(NO₃)₃. We also investigate the influence when using different concentration of H₂O₂. We did many analysis to proof our result, such as Field-emission scanning electron microscope(FE-SEM), Alpha step, X-ray diffraction(XRD), X-ray photoelectron spectrometer(XPS).

Experimental



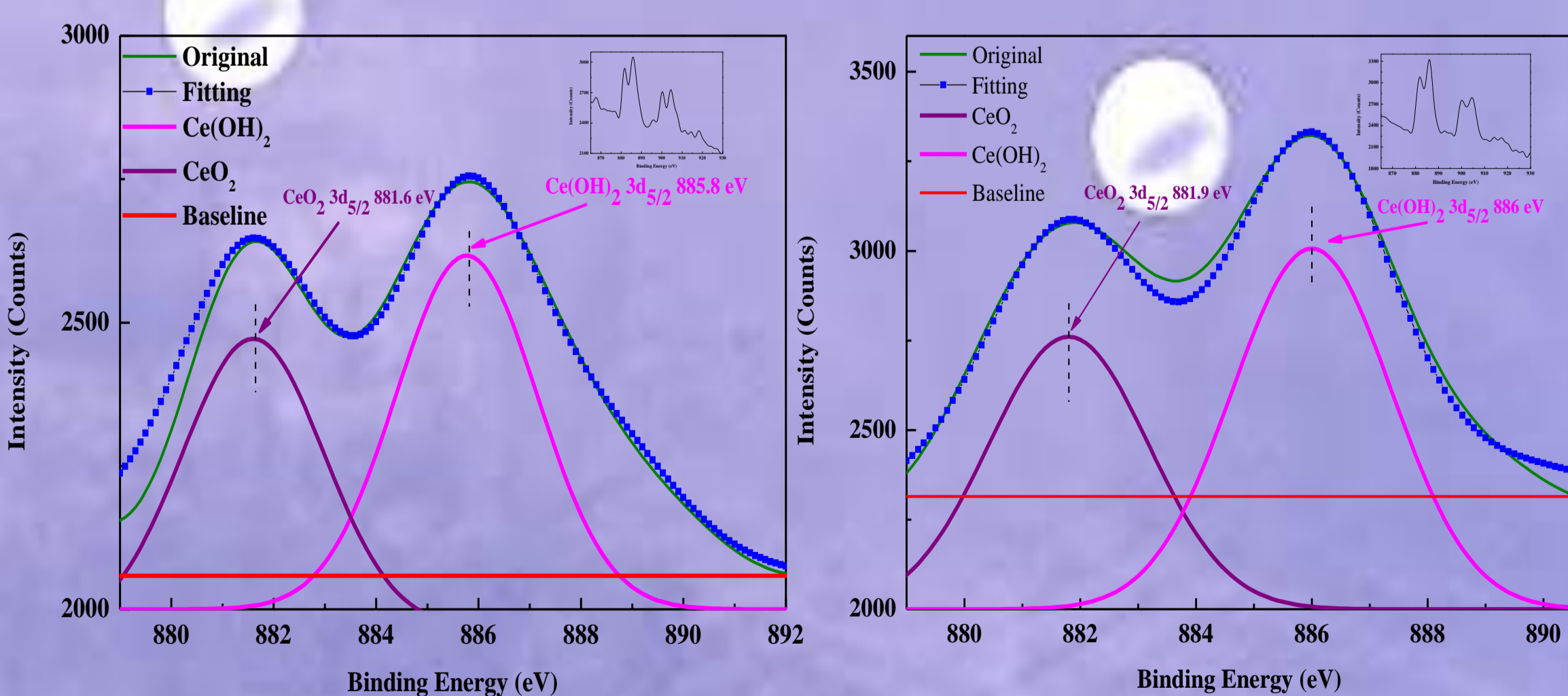
SEM images of (a)CeCl₃ and (b)Ce(NO₃)₃ doped ZnO nanorods

Result and discussion

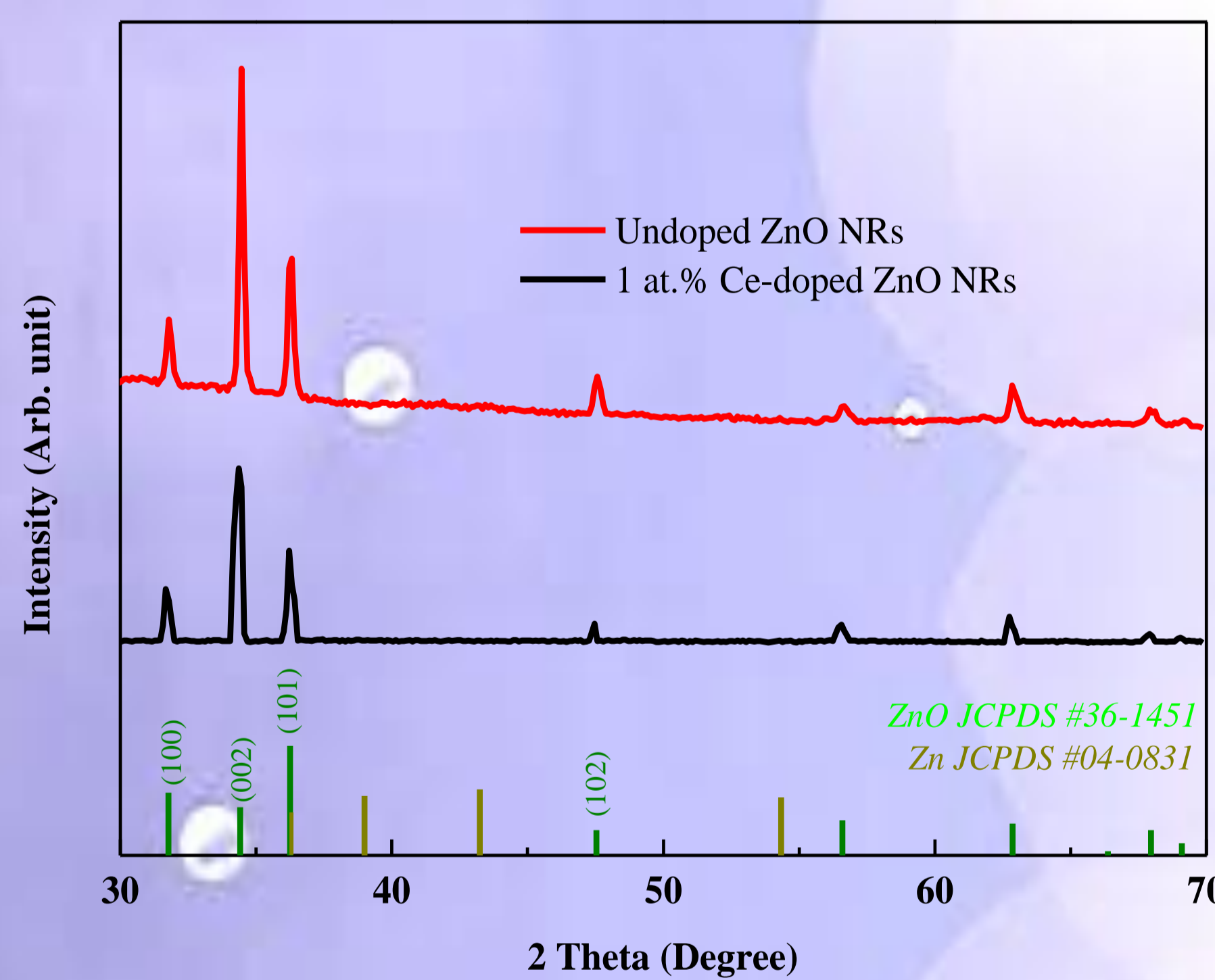


CeCl₃ and Ce(NO₃)₃ deposition

After achieving its maximum current, it start nucleating. And then cover ITO surface. Therefore the resistance become large, which cause current to decrease.



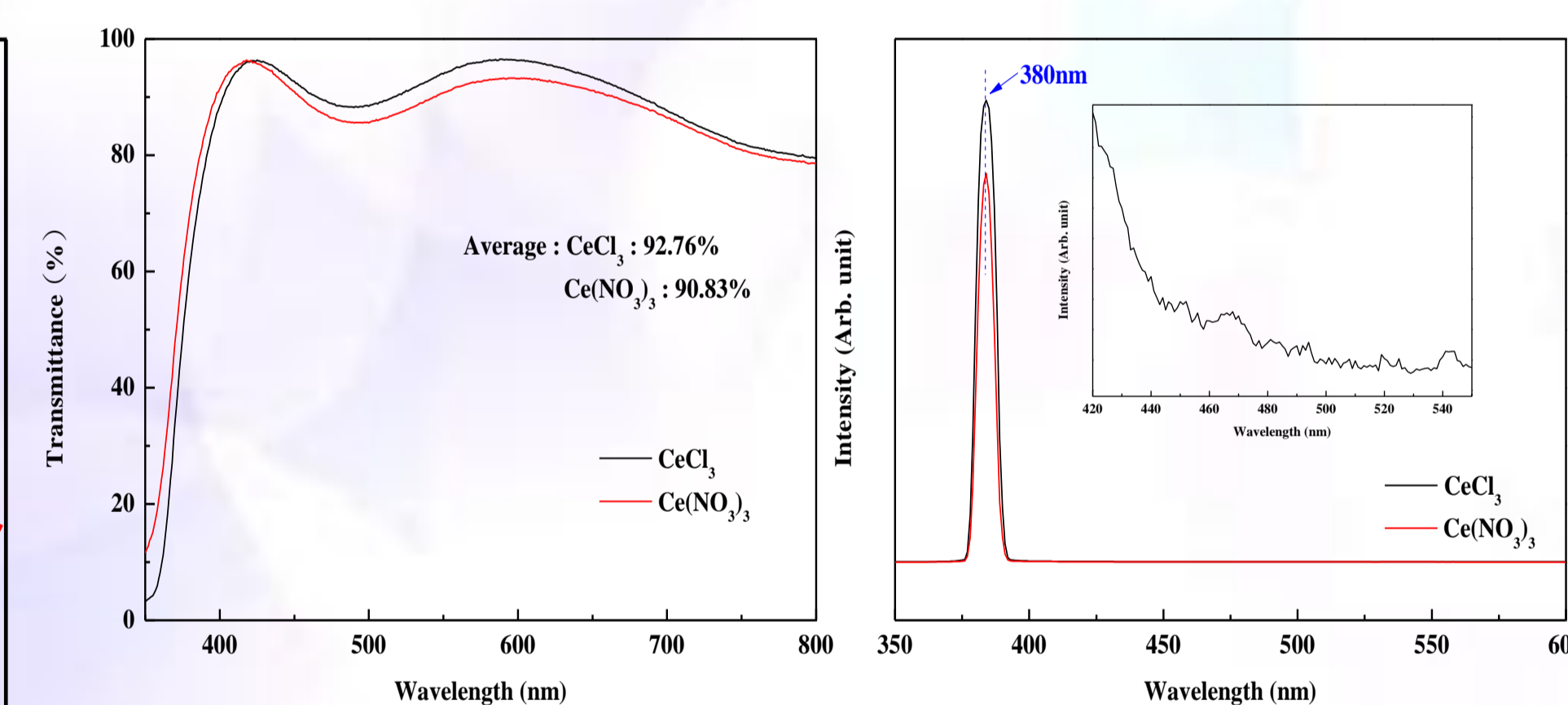
XPS images of (c)CeCl₃ and (d)Ce(NO₃)₃ doped ZnO nanorods



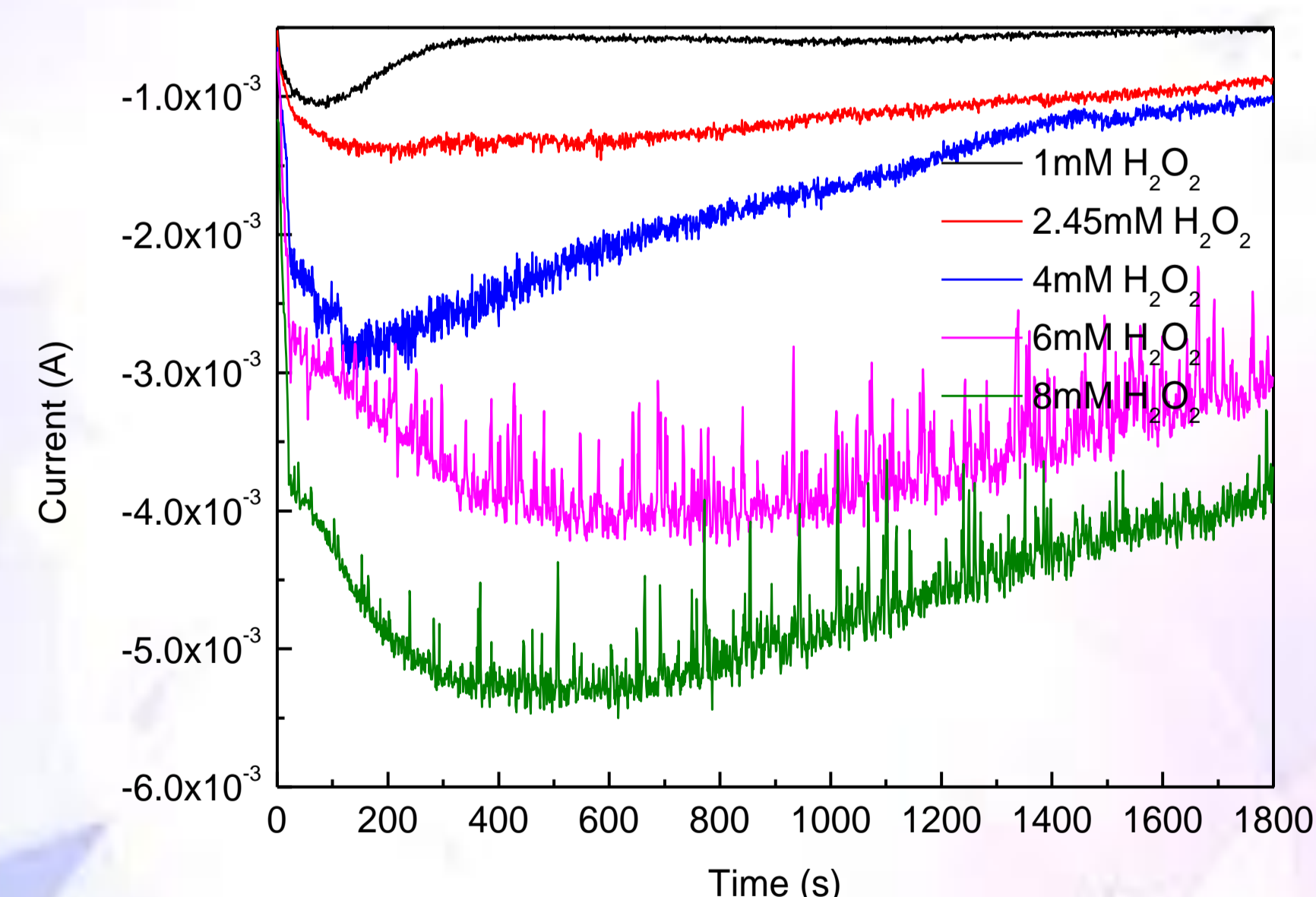
XRD image of undoped ZnO nanorods and Ce-doped ZnO nanorods.

This figure shows the preferred orientation is (002), and when we doped ZnO nanorodes with Ce, its intensity will decrease.

Ce 3d_{5/2} peak was located at 881.6 and 881.9 eV which indicates that this binding energy was attributed by Ce⁴⁺ ion doping.



Transmittance and PL images of Ce-doped ZnO nanorods



CeCl₃ Deposition of different concentration of H₂O₂

Due to electrochemical active substances, the current will increase with the increasing concentration of H₂O₂.

Reference

- [1]黃茂嘉, 「奈米氧化鋅結構之電化學研製及其在發光二極體之應用」, 中央大學, 碩士論文, 民國100年。
- [2]X.Y. Chen, F. Fang, A.M.C. Ng et al. *Thin Solid Films* **520**, 1125–1130, (2011)

Conclusion

- ✓ The transmittance exceed 90% which reach our goals.
- ✓ The best potential in our study to do cathode deposition is -1.0 V.
- ✓ Both Ce(NO₃)₃ and CeCl₃ are hexagonal wurtzite structure.
- ✓ There are good result of SEM when the concentration of H₂O₂ between 2.45 mM and 4 mM.